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Optimization of natural and synthetic suspending agents present in diclofenac suspension

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ABSTRACT

Suspensions are coarse dispersions defined as a biphasic liquid dosage form of medicaments in which insoluble solid particles are suspended uniformly in a liquid. The size of the insoluble solid particles in liquid phase may exceed 0.1µm.Dispersed phase is also known as internal phase and continuous phase is also known as external phase. Present research work focuses on effect of suspending agents in diclofenac suspension. Both natural and synthetic suspending agents are used. The natural suspending agent is gelatin and synthetic suspending agents of gelatin (2%, 3%, and 4%) and carbopol ((2%, 3%, and 4%) was prepared. Six formulations with formulation code of F1, F2, F3, F4, F5, and F6 was formulated. The prepared suspensions are evaluated for viscosity,pH, sedimentation volume, degree of flocculation, particle size etc.

INTRODUCTION

Suspensions are choice of formulation when the drug is insoluble in aqueous and non aqueous solvent. Suspensions prolong the duration of action and increase the stability of drug which is liable to hydrolysis. [1] Suspensions offer more advantages than solid dosage forms. They are easy to swallow higher bioavailability and prolong the duration of action. The type of suspensions includessuspensions for oral use, parenteral use and external use. Suspensions are classified as flocculated and deflocculated suspension. Suspending agents are used to improve physical

stability, and also enhance viscosity of the continuous phase by remaining as suspended particles in liquid for a long time. [2] Hydrocolloids are used for aqueous suspension, and sodium carboxy methyl cellulose was used for parenteral suspension.

The factors to be considered while selecting suspending agents are

- 1. Settling of particles
- 2. Pouring of suspension through the bottle
- 3. Spreading of suspension on the affected area of the skin

Table no 1: Explanation of ingredients used in the formulation

Some of the suspending agents are Natural like gelatin and sodium alginate and synthetic like carbopol, polyvinyl alcohol etc [3]

MATERIALS AND METHODS

All the chemicals used were of analytical grade. Chemicals used and their uses are shown in table no 1

		_		-			
	S.No	Ingredients	Compa	ny	Category		
	1	Diclofenac	Madras	Madras pharmaceuticals		Anti -inflammatory	
	2	Gelatin	Nice chemicals		Natural susp		
	3	Carbopol	Nice ch	Nice chemicals Nice chemicals		Synthetic suspending agent Buffer	
	4	Potassium chloride	Nice ch				
	5	Sucrose	Nice ch	emicals	Sweetening	agent	
		ŋ	Table no 2:	: Formulation Ta	ble		
No	Ingredients	F1	F2	F3	F4	F5	F6
	Diclofenac	1gm	1gm	1gm	1gm	1gm	1gm
	Gelatin	2gm	3gm	4gm	-	-	-
	Carbopol	-	-	-	2gm	3gm	4gm
	Potassium	1gm	1gm	1gm	1gm	1gm	1gm
	chloride						
	Sucrose	10mg	10mg	10mg	10mg	10mg	10mg
	Water	Q.S to	Q.S to	Q.S to	Q.S to	Q.S to	Q.S t
		100ml	100ml	100ml	100ml	100ml	100m

Method of preparation [4]

One gram of diclofenac was weighed and triturated in a mortar with a pestle. The suspending agents were added along with water, sucrose and potassium chloride. The slurry was transferred to 100ml measuring cylinder and make up the volume up to 100ml. The ingredients quantities are shown in table no 2.

Evaluation

> 5 6

Sedimentation volume [5]

Suspension of 100ml was stored in a 100ml measuring cylinder for 1hour at room temperature and observations were made for every 15 mins for 1 hour. The sedimentation volume was calculated using the following formula. Sedimentation volume (F) = Hu/Ho Ho = Initial height of suspension

Hu = Ultimate height of suspension

Degree of redispersibility [6]

Fixed volume of suspension 100ml was kept in a measuring cylinder which was stored at a room temperature for 1 hour at regular time intervals, measuring cylinder was moved upside, down until there was no sediment at the bottom of the cylinder.

Determination of pH [7]

Using pH meter the pH of diclofenac suspension was measured.

Viscosity determination

Using Brookfield viscometer the viscosity of the prepared suspension was measured. All the determinations were made in triplicate and results obtained were expressed in mean values.

Stability studies [8]

The shelf life of the product was fixed by performing stability studies for the prepared formulation. Accelerated stability studies was conducted for the prepared formulation by storing the containers at $40\pm2^{\circ}$ C temperature and $75\pm5\%$ RH and studied for one month.

RESULTS AND DISCUSSION

Sedimentation volume

Sedimentation volume (F) = Hu /Ho Ho = Initial height of suspension Hu = Ultimate height of suspension The sedimentation volume results are shown in table No 3

S.No	Formulation code	Sedimentation Volume
1	F1	0.45±0.2
2	F2	0.54±0.4
3	F3	0.65±0.1
4	F4	0.89±0.5
5	F5	0.98 ± 0.8
6	F6	0.97 ± 0.9

Table No 3Sedimentation Volume of the prepared Diclofenac Suspensi
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pH test

The pH values are recorded for all the six formulations. The F5 formulation shows 7.23 ± 0.3

pH which was nearer to the skin pH. The results are shown in table no 4.

Table N	ofenac Suspension		
	S.No	Formulation code	рН
	1	F1	8.12±1.2
	2	F2	8.53±0.8
	3	F3	8.28 ± 0.5
	4	F4	6.92±0.1
	5	F5	7.23±0.3
	6	F6	6.50±0.2

All the values are expressed as mean $\pm \overline{SD,n=3}$

Viscosity test

The	vis	cosity	test	was	perfo	ormed	using
Brookfie	ld	viscon	neter.	The	F3	form	ulation

containing both natural and synthetic humectant shows better viscosity than other formulations. The results are shown in table no 5

Table no 5 Viscosity of prepared formulations					
S.No	Formulation code	Viscosity			
1	F1	21.4±0.6			
2	F2	28.0±0.2			
3	F3	52.1±0.4			
4	F4	62.3±0.8			
5	F5	65.5±0.2			
6	F6	$122.0{\pm}~0.5$			

All the values are expressed as mean ±SD,n=3

From the viscosity studies it was concluded that F5 formulation shows moderate viscous than other formulations, easily pourable from the container, and it has ability to retain in the skin for a long time.

Stability Studies

The stability study was conducted for the selected formulation F5 as per ICH guidelines for a period of 1 month. The stability results are shown in table no 6

S.No	Parameters	Initial	Final			
			1 st week	2 nd	3 rd	4 th week
				Week	Week	
1	Color	White	White	White	White	White
2	pН	7.23±0.3	7.1 4±0.1	7.00 ± 0.4	7.28±0.3	7.11±0.5
3	Viscosity	65.5 ± 0.2	63.2±0.3	64.5±0.1	65.2±0.4	64.5±0.6
	(Centipoise)					
4	Sedimentation	0.98±0.2	0.97 ± 1.2	0.98±0.4	0.96±0.1	0.99±1.3
	Volume					
5	Degree of re	Readily	Readily re	Readily	Readily	Readily
	dispersibility	redispersible	dispersible	redispersible	redispersible	redispersible

All the values are expressed as mean ±SD,n=3

From the results it was found that the selected formulation F5 was found to be stable. There was no significant change from initial readings to final results after 1 month of stability studies.

CONCLUSION

The suspension was prepared by trituration method. F5 formulation was found to be best when compared to other formulations. The pH was found to be 7.23 ± 0.3 which are nearer to skin pH and viscosity was found to be 65.5 ± 0.2 . The sedimentation volume of the selected formulation was found to be 0.98 ± 0.2 which is near to 1

indicated good suspend ability where as other formulations are not near to 1.From the stability studies it was confirmed that the prepared formulation F5 was found to be stable. So the formulation F5 containing carbopol 3% is effective in all the aspects.

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