

# INTERNATIONAL JOURNAL OF PHARMACY AND ANALYTICAL RESEARCH

## ISSN: 2320-2831

IJPAR /Vol.11 / Issue 2 / Apr - Jun -2022 Journal Home page: www.ijpar.com

Research article

**Open Access** 

## DEVELOPMENT AND VALIDATION OF A HPLC METHOD FOR THE DETERMINATION OF ANTIDIABETICS (CANAGLIFLOZIN AND METFORMIN) IN API AND PHARMACEUTICAL DOSAGE

\*Ansari Abdul Mutalib, \*H. Parameshwar, \*A.V. Jithan, \*Sai Lakshmi, \*C.A Sri Ranjani, \*K. Sindoora

\*Omega College of Pharmacy, Edulabad, Ghatkesar, Affiliated to Osmania University, Hyderabad, Telangana.

Corresonding Author: Dr. H.Parameshwar Email: <u>parampharma@gmail.com</u> Email: <u>mutallib896@gmail.com</u>

## ABSTRACT

A sensitive & selective stability indicting RP-HPLC method has been developed & validated for the analysis of Canagliflozin & Metformin API. Based on peak purity results, obtained from the analysis of samples using described method, it can be concluded that the absence of co-eluting peak along with the main peak of Canagliflozin & Metformin indicated that the developed method is specific for the estimation of Canagliflozin & Metformin. Further the proposed RP-HPLC method has excellent sensitivity, precision and reproducibility.

-----

Keywords: RP-HPLC, Canagliflozin, Metformin, method development, validation parameters

## **INTRODUCTION**[<sup>1,2,3</sup>]

The development of any new or improved method usually tailors existing approaches and instrumentation to the current analyte, as well as to the final needs or requirements of the method. Method development usually requires selecting the method requirements and deciding on what type of instrumentation to utilize and why.

There are several valid reasons for developing new methods of analysis:

- There may not be a suitable method for a particular analyte in the specific sample matrix.
- Existing methods may be too error, artifact, and/or contamination-prone, or they may be unreliable (have poor accuracy or precision).

Reversed phase mode is the most popular mode for analytical and preparative separations of compound of interest in chemical, biological, pharmaceutical, food and biomedical sciences. In this mode, the stationary phase is non polar hydrophobic packing with octyl or octa decyl functional group bonded to silica gel and the mobile phase is polar solvent. An aqueous mobile phase allows the use of secondary solute chemical equilibrium (such as ionization control, ion suppression, ion pairing and complexation) to control retention and selectivity. The polar compound gets eluted first in this mode and non polar compounds are retained for longer time. As most of the drugs and pharmaceuticals are polar in nature, they are not retained for longer times and hence elute faster. The different columns used are octa decyl silane (ODS) or  $C_{18}$ ,  $C_8$ ,  $C_4$ , etc., (in the order of increasing polarity of the stationary phase).

There may be a need for an alternative method to confirm, for legal or scientific reasons, analytical data originally obtained by existing methods.

#### DRUG PROFILE CANAGLIFLOZIN

**Description:** Canagliflozin belongs to a new class of anti-diabetic drugs that works by inhibiting the sodium-glucose transport protein (SGLT2). This transport protein is found in the kidney and is responsible for reabsorbing glucose that has been filtered. FDA approved on March 29, 2013.

Structure



Chemical formula
Molecular mass
Physical appearance
Solubility

: C<sub>24</sub>H<sub>25</sub>FO<sub>5</sub>S : 444.516 g/mol

: Canagliflozin hemihydrate drug substance is a white to off white powder

: It is Soluble in DMSO (88 mg/ml at 25 °C), water (<1 mg/ml at 25 °C), ethanol (<1 mg/ml at 25 °C), and methanol.

: Antidiabetic drug.

#### Category Machanism of action

**Mechanism of action** : Sodium-glucose co-transporter 2 (SGLT2), expressed in the proximal renal tubules, is responsible for the majority of the reabsorption of filtered glucose from the tubular lumen. Canagliflozin is an inhibitor of SGLT2. By inhibiting SGLT2, canagliflozin reduces reabsorption of filtered glucose and lowers the renal threshold for glucose (RTG), and thereby increases urinary glucose excretion.

#### **METFORMIN HCL**

Structure:



**Chemical Formula**: C<sub>4</sub>H<sub>11</sub>N<sub>5</sub>

**Description:** This compound belongs to the class of organic compounds known as biguanides. These are organic compounds containing two N-linked guanidines.

**Solubility**: Slightly soluble in ethanol and also slightly soluble in alcohol, and insoluble in substances such as acetone, ether, and chloroform.

**Indication:** For use as an adjunct to diet and exercise in adult patients (18 years and older) with NIDDM. May also be used for the management of metabolic and reproductive abnormalities associated with polycystic ovary syndrome (PCOS).

#### MATERIALS AND INSTRUMENTS

**Drug Sample:** Were obtained from Canagliflozin and .Metformin - Syncorp Clincare Technologies Pvt. Ltd., Hyderabad.

**Chemicals and Solvents Used:**1.Water –HPLC grade, 2. Methanol – HPLC grade 3. Acetonitrile - HPLC grade, 4. Tri Ethyl Amine -HPLC grade, 5. Potassium Dihydrogen Phosphate, 6. Dipotassium Hydrogen Phosphate

All the above chemicals and solvents were supplied by S.D. Fine Chemicals Ltd., India; Qualigens Fine Chemicals Ltd., Mumbai, India and Ranbaxy Chemicals Ltd., New Delhi, India.

**Instruments Used:** 1. Shimadzu ATY224 Digital Electronic Balance 2. Wensar Pvt. Limited, India, pH meter, 3. Vacuum Pump Filter (PCI) 4. Life Care Sonicator5. Elico SL 210 Double beam UV/Vis Spectrophotometer 6. Hitachi Elite Lachome HP LC model Containing Pump: L2130, Detector: UV, L2400, Software: D2000 Elite.

#### **METHOD DEVELOPMENT**

**HPLC Method Development**: The parameters for the development were as follows:

1) Selection of detection Wavelength, 2) Selection and optimization of mobile phase, 3) Study of system suitability parameters, 4) Application of the method to the drug in pure form.

Validation Of Proposed Method: The parameters for the validation were carried out as follows:

System suitability 2) Linearity 3) Accuracy 4) Precision
Robustness 6) Ruggedness 7) Limit of detection 8) Limit of quantitation

*Selection of chromatographic method for separation* Reverse phase chromatographic technique was selected since the drug is polar in nature.

#### HPLC METHOD DEVELOPMENT Selection of wavelength

Selectivity of HPLC method that uses UV detector depends on proper selection of Wavelength. A wavelength which gives good response for the drug to be detected should be selected. From the UV spectra 238 nm was selected as the wavelength for Metformin & 275nm for Canagliflozin. The  $\lambda$  max selected for work was 238nm.



Fig 1: UV-Spectrum for Metformin (238nm)



Fig 2: UV-Spectrum for Canagliflozin (275nm)

#### Selection of Mobile phase

Solvent selectivity (solvent type), solvent strength (percentage of organic solvent in the mobile phase), From the various mobile phase tried, mobile phase containing 0.01M Phosphate buffer (pH-3.6): Methanol (35:65) % v/v was selected, since it gave sharp peak with symmetry within limits and significant retention time.

#### **Reagent and Pharmaceutical preparations**

An analytically pure sample of Canagliflozin and Metformin was procured as gift sample from Syncorp Clincare Technologies Pvt. Ltd. (Hyderabad, India). HPLC grade Acetonitrile and HPLC grade Water procured from E. Merck (Ahmadabad). Orthophosphoric acid and Tri ethyl amine (AR grade, purity 99.5 %), dihydrogen potassium phosphate and Dipotassium hydrogen phosphate (AR grade) was procured from Qualigens.

#### **Preparation of Phosphate buffer**

About 1.36089 grams of Potassium dihydrogen orthophosphate was weighed and transferred into a 1000ml beaker, dissolved and diluted to 1000ml with HPLC water. The pH was adjusted to 3.60 with diluted Orthophosphoric acid.

#### Preparation of mobile phase

650ml of Methanol and 350ml of Buffer are mixed. The solution was degassed in an ultrasonic water bath for 10 minutes and filtered through  $0.45 \mu$  filter under vacuum.

#### **Preparation of stock solution**

A stock solution of Canagliflozin was prepared by accurately weighing 25 mg of drug, transferring to 25 ml volumetric flask, Add about 15 ml of mobile phase and sonicated to dissolve it completely and make up volume up to mark with mobile phase ( $1000\mu g/ml$ ). A stock solution of Metformin was prepared by accurately weighing 25 mg of drug, transferring to 25 ml volumetric flask, Add about 15 ml of mobile phase and sonicated to dissolve it completely and make up volume up to mark with mobile phase ( $1000\mu g/ml$ ). From above stock solution final concentrations should be prepared. The final concentrations for Canagliflozin is 2 ppm & Metformin for 20 ppm

#### Chromatographic experiment

Different chromatographic conditions were tried to optimize the method, optimized method is following:

I dole	i opunizeu en oniutogrupine conditions
Stationary phase	C <sub>18</sub> Hypersil, 150 x 4.6, particle size 3µ
Injector	Rheodyne
Mobile phase	Phosphate Buffer : Methanol (60:40% v/v) pH: 3.6
Flow rate	1ml/min
Injection volume	20µl
Wavelength	257nm
Temperature	250C
Run time	10 min

#### **Table 1. Optimized Chromatographic Conditions**



Fig 3: Chromatogram for Blank Solution



Fig 4: Chromatogram for Optimized Chromatographic Method



206

#### Assay of Metformin & Canagliflozin In Pharmaceutical Dosage Form

Twenty tablets were taken and the I.P. method was followed to determine the average weight. Above weighed tablets were finally powdered and triturated well. A quantity of powder equivalent to 100 mg of drugs were transferred to 100 ml volumetric flask, and 70 ml of mobile phase was added and solution was sonicated for 15 minutes, there after volume was made up to 100 ml with same solvent. Then 10 ml of the above solution was diluted to 100 ml with mobile phase. The solution was filtered through a membrane filter  $(0.45 \ \mu\text{m})$  and sonicated to degas. From this stock solution  $(3.5 \ \text{m})$  was transferred to five different 10 ml volumetric flasks and volume was made up to 10 ml with same solvent system. The solution prepared was injected in five replicates into the HPLC system and the observations were recorded. A duplicate injection of the standard solution was also injected into the HPLC system and the peak areas were recorded. The data are shown in Table 2.

	T	abl	e 2:	Recovery	Data for	estimation	Metformin	and	Canagliflozin
--	---	-----	------	----------	----------	------------	-----------	-----	---------------

Brand name of Tablets (Invokamet Tab (Janssen Pharmaceuticals, Inc.)	Labelled amount of Drug (mg)	Mean (±SD) amount (mg) found by the proposed method (n=6)	Assay + % RSD
Canagliflozin	50	49.895 (± 0.07)	99.864 (± 0.49)
Metformin	500	499.974 (± 0.07)	99.876 (± 0.46)

The amount of drugs in Invokamet tablet was found to be  $99.876 (\pm 0.46)$  mg/tab for Metformin and  $99.864 (\pm 0.49)$  mg/tab for Canagliflozin.



Fig 8: Sample Chromatogram-2

#### **STABILITY STUDIES**

Following protocol was strictly adhered to for forced degradation of Canagliflozin and Metformin Active Pharmaceutical Ingredient (API). The API (Canagliflozin and Metformin) was subjected to stress conditions in various ways to observe the rate and extent of degradation that is likely to occur in the course of storage and/or after administration to body. This is one type of accelerated stability studies that helps us determining the fate of the drug that is likely to happen after along time storage, within a very short time as compare to the real time or long term

stability testing. The various degradation pathways studied are acid hydrolysis, basic hydrolysis, thermal degradation, photolytic degradation and oxidative degradation.

#### **Results of forced degradation studies**

The results of the stress studies indicated the specificity of the method that has been developed. Canagliflozin and Metformin were stable in oxidation and thermal stress (wet heat) conditions. The result of forced degradation studies are given in the following Table 3.

Table 5. Results of forced degradation studies of Canaginozin and Methorinin ATT.						
Stress condition	Time	Assay of active substance	Assay of degraded products	Mass Balance (%)		
Acid Hydrolysis (0.1 M HCl)	24Hrs.	86.44	13.56	100.0		
Basic Hydrolysis (0.I M NaOH)	24Hrs.	97.37	2.63	100.0		
Wet heat	24Hrs.	87.06	12.94	100.0		
UV (254nm)	24Hrs.	61.33	38.67	100.0		
3 % Hydrogen peroxide	24Hrs.	89.07	10.93	100.0		

#### Table 3: Results of forced degradation studies of Canagliflozin and Metformin APL

### **RESULT & DISCUSSION**

To develop a precise, linear, specific RP-HPLC method for analysis of Canagliflozin and Metformin different chromatographic conditions were applied & the results observed are presented in the thesis. In case of RP-HPLC various columns are available, but here  $C_{18}$  Hypersil, 150 x 4.6, particle size 3µ column was preferred because using this column peak shape, resolution and absorbance were good. Mobile phase & diluent for preparation of various samples were finalized after studying the solubility of API in different solvents of our disposal (methanol, acetonitrile, dichloromethane, water, 0.1N NaOH, 0.1NHCl). Canagliflozin was found to be Soluble in DMSO (88 mg/ml at 25 °C), water (<1 mg/ml at 25 °C), ethanol (<1 mg/ml at 25 °C), and methanol. Metformin was found to be freely soluble in water, Methanol slightly soluble in ethanol and also slightly soluble in alcohol, and insoluble in substances such as acetone, ether, and chloroform. Detection wavelength was selected after scanning the standard solution of drug over 200 to 800nm. From the U.V spectrum of

#### REFERENCES

- 1. Instrumental methods of chemical analysis. by B.K. Sharma, pp. 75-8, 113-5.
- 2. Instrumental methods of chemical analysis. 5th ed, by Galen W. Ewing. p.1.
- 3. Brochmann E, Hanssen H. Pharmaceutical analysis. 1st ed, by Takeru Higuchi. p. 1-10.
- 4. Beckett JB Stenlake. Practical pharmaceutical chemistry. 4th ed. Vol. II, by A.H. p. 275-98.
- 5. Quality assurence, worth the effort, Inforum, october 2003 volume 7. Vol. 4.
- 6. Sethi. Quantitative Analysis of drugs in Pharmaceutical formulation. 3rd ed, by P.D. p. 1-21, 51-6.
- 7. Kasture et al. Hand book of Pharmaceutical Analysis, Vol 1. High Performance Liquid Chromatography, 2001. pl1.
- 8. Validation of analytical procedures, methodology, ICH harmonized tripartite guidelines; 1996.
- 9. Text on validation of analytical procedures, ICH harmonized tripartite guidelines; 1994.
- 10. Shankar R. A Text book of Pharmaceutical Analysis. 3rd ed. p. 2.2.
- 11. Lacy CF, Armstrong LL, Goldman MP, Lance LL. Lexi-Comp's drug information handbook. 12th ed. Lexi-Comp Inc. ISBN 1-59195-083-X; 2004.
- 12. Momin MY, Yeole PG. Puranik MP Institute of pharmaceutical Education and Research, Borgan(Meghe). India: Wardhan. Available from: http://www.ijpsonline.com/text.asp?2006/68/3/387/26672.
- 13. Practical HPLC method development by Snyder. Glajch and Kirkland, A. Wiley. Interscience Publication, p. 4-10, 92-102.
- 14. Bioanalytical method validation. United States Department of Health and Human Services, Food and Drug Administration, Center for Drug Evaluation and Research (CDER), Center for Veterinary Medicine (CVM); May 2001.
- 15. Center for Drug Evaluation and Research (CDER).Reviewer Guidance. ICH, Q1A (R2) stability testing of new drug substances and products (Geneva, Feb. 2003). Validation Chromatogr Methods. Nov 1994.

Canagliflozin and Metformin it is evident that most of the HPLC work can be accomplished in the wavelength range of 215-290 nm conveniently. Further, a flow rate of 1.0 ml/min & an injection volume of 20  $\mu$ l were found to be the best analysis. The result shows the developed method is yet another suitable method for assay which can help in the analysis of Canagliflozin and Metformin in different formulations.

#### CONCLUSION

A sensitive & selective stability indicting RP-HPLC method has been developed & validated for the analysis of Canagliflozin & Metformin API. Based on peak purity results, obtained from the analysis of samples using described method, it can be concluded that the absence of co-eluting peak along with the main peak of Canagliflozin & Metformin indicated that the developed method is specific for the estimation of Canagliflozin & Metformin. Further the proposed RP-HPLC method has excellent sensitivity, precision and reproducibility.