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**Research article** 

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# Development and validation of the quantitative analysis of amlodipine besylate in tablet formulation by fourier transform infrared spectroscopy

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# ABSTRACT

A simple, precise, non destructive, rapid, and inexpensive, ecofriendly and reproducible FTIR spectrophotometric method for quantitative determination of amlodipine besylate in tablet formulation was developed and validated. This method concerned with absorbance measurements of bands corresponding to c=0 stretch centered by 1731-1687 cm<sup>-1</sup>. Analytical method validation was carried out to study the parameters as linearity, repeatability, precision, accuracy as well as photo degradation study. The linearity range was found to be 0.3-1.8 % w/w (regression equation: Y=0.585x-0.044, r<sup>2</sup> = 0.994). The data show good precision results of this method, since the RSD values observed less than 2 % as per ICH limits. The proposed FTIR method was successfully applied to the assay of amlodipine besylate in tablet formulation.

**Keywords:** Fourier Transform Infrared Spectroscopy, Quantitative Determination, Quality Control, Amlodipine Besylate.

# **INTRODUCTION**

Chemically, Amlodipine Besylate (AB) (CAS No. 111470-99-6), 3-ethyl-5-methyl (4RS)-2-[(2-aminoethoxy) methyl]-4-(2-chloro phenyl)-6-methyl-1, 4-dihydropyridine-3,5-dicaboxylate benzene sulphonate(empirical formula  $C_{20}H_{25}CIN_2O_5.C_6H_6SO_3$ ) is a second generation 1,4-dihydropyridine derivative of the prototypical molecule Nifedipine and used in treatment of chronic stable angina and in the management of

mild to moderate essential hypertension as antihypertensive.<sup>[1]</sup>

Many methods reported in literature for determination as UV spectrophotometric method<sup>[2-24]</sup>, UV-RP-HPLC<sup>[25]</sup>, UV-TLC<sup>[26]</sup>, UV-HPTLC<sup>[27]</sup>, RP-HPLC<sup>[28-36]</sup>, Stability indicating HPLC<sup>[37,38]</sup>, TLC-HPTLC<sup>[39]</sup>, LC-MS/MS<sup>[40]</sup>, Stability indicating LC<sup>[41]</sup> and FTIR-Raman-NMR<sup>[42]</sup>, but no such method available for its determination using Fourier Transform Infrared Spectroscopy. Hence the endeavor was made to develop ecofriendly,

green, cost effective, simple, rapid and non destructive method using FTIR for the assay of

Amlodipine Besylate in pure and tablet forms.



Figure 1: Chemical structure of amlodipine besylate

# **EXPERIMENTAL**

#### **Chemicals and reagents**

Standard sample of AB was obtained as gift from Lupin Limited, Aurangabad, and (MH) India. For recording spectra, IR spectroscopic grade of KBr obtained from Merck KGaA, Germany. Single dosage form and combination tablets (5mg dose) of different brands were procured from local market of Nanded, (MH) India.

#### **FTIR Instrumentation**

FTIR analyses were carried out on IR Affinity-1 (00722) FTIR spectrophotometer, (Shimadzu, Japan) with IR Solution 1.50 version of software for data analysis DLATGS detector with ceramic light source of high luminance. KBr disc spectra recorded in mid IR region 4000cm<sup>-1</sup> and 400cm<sup>-1</sup>, with average 45 scans having resolution of 4cm<sup>-1</sup>. To carry out the forced photo degradation study, photochemical reactor (Shrinivasan Griffin-Rayonet type) was used.

#### **Calibration curve**

Different six concentrations of AB individually prepared in range of 0.3-1.8% w/w for calibration curve. Specific quantity of API diluted with 1000mg KBr and triturated by maintaining homogeneity.

## **Recovery Efficiency and Method validation**

Developed method was validated for its accuracy by addition of known concentration of API to pharmaceutical sample (80, 100, 120% w/w) then total concentration estimatated by developed method. Reproducibility studied for pure drug in set of sixes and interferences studies for excipients with API in tablet dosage form also carried out. The API and excipients extracted from tablets with help of chloroform. The developed method indicates no interference of excipients in tablet formulation with pure drug.

#### Accuracy

Standard addition method was applied with recovery of pure drug from excipients at three different quantities (80, 100, 120% w/w) to the tablet powder of Amlogold-5 (pure AB) and Amlopress L 5/5 (combination) for each tablet as 1% w/w dilution into potassium bromide in triplet. The sample homogeneity maintained.

#### Linearity

Linearity of the method studied for six samples of different concentration range 0.3-1.8% w/w. Calibration curve linearity assessed by the linear regression line equation for both API.

#### Precision

Repeatability for API (In single of each) was studied with pure API at 1% w/w concentration six times on same day (Intraday). Intermediate precision assessed for precision of pharmaceutical formulations on interday.

#### **Analysis of Marketed Tablet Formulations**

Five different brands of single dose of AB (Amlogold-5, Amlovas-5, Amodep, Amloz, and Amlokind) were used for estimation of drug content. 20 tablets were weighed and average

weight determined then triturated to fine powder. Appropriate amount of each tablet diluted to potassium bromide for getting 1% w/w concentration. Readings were taken in triplets.

#### **Interference study**

Literature revealed that no such method reported for AB. No sample pretreatment, no use of hazardous chemicals, less time consuming with no requirement of special efficiency. The effect of other components commonly excipients analyzed in pharmaceutical formulation for these studies. Excipients and API separated from marketed preparation of different brands mentioned above with chloroform and spectra were taken. AB and its standard had no interference with absorption bands present in mixture. Fig. 2.

#### **RESULT AND DISCUSSION**

The FTIR spectrum for pure sample of amlodipine besylate exhibited absorbance bands in the range of 750, 1715, 1650, 1697, 1735, 1375, 3500-3100, 1100, 3500-3100, 1600-1450, and 1300-1000 cm<sup>-1</sup>. The low-intensity absorbance bands arising from AB was not much affected by dilution in dry potassium bromide; therefore, in the present study we have used dry potassium bromide as the diluent.

The most prominent absorbance band corresponding to the carbonyl group centered in the range of  $1731-1687 \text{ cm}^{-1}$  for the diluted samples of AB in dry potassium bromide was within the 2.0 absorbance units. The absorbance spectra for the

diluted AB samples of various concentrations are shown in Figure 2a-2f.

The peak height (intensity) for the peak centered in the range of 1731-1687 cm<sup>-1</sup> for AB was used for the preparation of calibration curve as shown in Figure 4. The calibration curve is described by the equation y = c + bx, where y represents peak area and x represents concentration of drugs (AB). Initially the samples in the concentration range of 0.5–5.0% w/w, 1-6% w/w, 2-24% w/w were analyzed to determine the linearity.

The calibration curve with good linearity was established ranging from 0.3 to 1.8% w/w for amlodipine besylate in potassium bromide. The corresponding linear regression equation for AB was y = -0.099+0.585x and the correlation coefficient for calibration curve was 0.994 for each drug (AB) (Table 1).

The FTIR spectrum in Figure 3 and 4 indicates that there is no interference of excipients used in the formulation of tablet dosage form.

Five different brands of amlodipine besylate (Figure 5, Table 10, 11) were analyzed using the developed method. The average recoveries of AB and their all the formulations were in the range of not less than 90 % and not more than 110 % as per USP acceptable limits and the % RSD values were in the range of 0.2 -0.5 (i.e. < 2) respectively, The % recovery of label claim was in good agreement and within the acceptable limits of the USP (not less than 90.0 % and not more than 110.0 % of the stated amount of AB)



Figure 1: FTIR Spectrum of Amlodipine Besylate (Standard)



Figure 2 a) : FTIR Spectrum of Amlodipine Besylate (0.3% w/w)



Figure 2 b) : Linearity spectrum of AB (0.6 % w/w)



Figure 2 c) : Linearity spectrum of AB (0.9 % w/w)



Figure 2 d) : Linearity spectrum of AB (1.2% w/w)



Figure 2) e : Linearity spectrum of AB (1.5 % w/w)







Figure 3) : FTIR spectrum of tablet excipients



Figure 4 : Excipients and AB (pure) overlay



Figure 5: FTIR spectra for Amlogold-5 tablet (5 mg label claim)



Figure 6: FTIR spectra for Amodep-5 tablet (5 mg label claim)



Figure 7 : FTIR spectra for Amlokind-5 tablet (5 mg label claim)







Figure 9: FTIR spectra for Amloz-5 tablet (5 mg label claim)

## Table 1: Linear Regression Data For Amlodipine Besylate

Parameters	AB
Wave number (cm <sup>-1</sup> )	1697 cm <sup>-1</sup>
Linearity range (% w/w)	0.3-1.8%
Limit of detection (% w/w)	0.0059
Limit of quantitation	0.01789
(% w/w)	
y = mx + c	Y=0.585x-0.044
Slope (m)	0.585
Intercept (c)	0.044
Regression coefficient R <sup>2</sup>	0.994

1 abic 2. Statistical valuation, analysis of tablet for mulation
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Name of the drug	Mean*	SD*	% RSD*
AB	99.86	0.242212	0.242535

\* Indicates average of six determinations

Tab	le 3	: Linear	regress	ion data i	for	calibi	ration	curve o	f AB

Name of the drug	Linearity range (% w/w)	$\mathbf{R}^2$	Slope	Intercept
AB	0.3-1.8 %	0.994	0.585	-0.099

Table 4 •	Precision	data of	marketed	formulation	(Amlogold_5)
Table 4.	I I CUSION	uata UI	mai keteu	101 mulation	(Annogoiu-5)

Sr. No.	Interval of Time	Concentration (% w/w)	% Recovery
		AB	AB
Ι		5	100.17
II	Day 1	5	100.52
III		5	101.13
Ι		5	100.34
II	Day 3	5	100
III		5	100.52

Table 5 : Statistical	validation	of intra-day	precision	data
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Name of the drug	Mean*	SD*	% RSD*
AB	100.60	0.485	0.482

\* Indicates average of six determinations

Table 6 : Statistica	ision data			
Name of the drug	Mean*	SD*	% RSD*	
AB	100.286	0.264	0.263	

\* Indicates average of six determinations

Table 7: Statistical data for repeatability of A					
Name of the drug	Mean*	SD*	% RSD*		
AB	99.93	0.282518	0.282693		

\* Indicates average of six determinations

Level of Recovery	Amount present	Added concentration (mg)	Amount recovered	% Recovery
	( <b>mg</b> )		( <b>mg</b> )	
	AB	AB	AB	AB
	5	4	9.09	101.04
80%	5	4	9.04	100.04
	5	4	9.03	100.34
	5	5	10.11	101.12
100%	5	5	10.06	100.69
	5	5	10.12	101.22
	5	6	11.09	100.87
120%	5	6	11.13	101.22
	5	6	11.11	101.04

Table 8:	Recovery	study d	lata
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Table 9: Statistical validation of recovery study data				
Level of Recovery	% Mean Recovery *	SD*	% RSD*	
	AB	AB	AB	
80%	101.80	0.0321	0.355	
100%	101.01	0.0321	0.318	
120%	101.04	0.02	0.180	

\* Indicates average of three determinations

Table 10: LOD & LOQ					
Name of the drug	LOD (% w/w)	LOQ (% w/w)			
AB	0.059	0.017			

Table 11: Statistical data for analysis of different brands of AB							
BRANDS	% RECOVERY			% LABEL CLAIM			
	Mean	SD*	%RSD*	Mean	SD*	%RSD*	
AMODEP-5	101.13	0.057	0.0570	101.10	0.10	0.098	
AMLOZ-5	101.11	0.011	0.0114	101.03	0.057	0.057	
AMLOVAS-5	101.16	0.011	0.0114	101.13	0.115	0.114	
AMLOKIND-5	100.67	0.020	0.0206	100.56	0.057	0.057	

## CONCLUSION

The proposed method is inexpensive and environmental friendly method. It eliminates complexity of usual extraction methods allowing faster analysis without using hazardous organic chemicals in accordance with the green chemistry needs and fulfils industrial demand of rapid and economical method. FTIR spectroscopy is employed for the qualitative analysis of pharmaceuticals; with advent in sampling techniques, DRIFT spectroscopy may serve as useful technique for qualitative and quantitative analysis of solid-state pharmaceuticals. The development and validation of eco-friendly FTIR method for the quantitation of solid-state API and its successful application to pharmaceuticals. This new approach of using transmission FTIR for direct determination of amlodipine besylate in pharmaceutical preparations where several excipients are present provides an alternate to expensive and time consuming procedure. The proposed FTIR method is simple, accurate, precise, validate ecofriendly and reproducible. It can be used for referring quality control analysis of amlodipine besylate in pharmaceutical dosage form.

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#### **CONFLICT OF INTERESTS**

The authors declared that there is no conflict of interests in the publication of the paper.

#### REFERENCES

- [1]. Wilson, C. O., Gisvold, O., Cardiovascular agents, *Textbook of Organic Medicinal and Pharmaceutical Chemistry*, Block, J. H., Beale, J. M., Philadelphia, Lippincott Williams and Wilkins, 1.1, 2004, 19.
- [2]. Mishra, P., Gupta, A., and Shah, K., Indian J. Pharm. Sci., 69, 6, 831-833.
- [3]. Bhatia, M.S., Kumbhar, S. T., Jadhav, S. D., and Bhatia, N. M., *Int. J. Pharm. Pharm. Sci.*, 3(4), 2011, 195-197.
- [4]. Patil, P. R., Rakesh, S. U., Dhabale, P. N., andBurade, K. B., *Res. J. Pharm. Technol.* 2(2), 2009, 304-307.
- [5]. Chaudhari, B. G., Int. J. Pharm. Chem. Biol. Sci., 2(4), 2012, 626-632.
- [6]. Misra, M., Kavathia, A., and Suchak. K., Int. J. Pharm. Pharm. Sci., 5(3), 2013.
- [7]. Hassan, S. A., Darwish, H. W., Salem, M. Y., and El-Zeany, B. A., Int. J. Spectrosc., 2013, Article ID 273102, 1-8. doi.org/10.1155/2013/273102
- [8]. Basnet, N. S., and Aryal, S., *Acta Pharm*, 58, 2008, 299–308. Doi:10.2478/v10007-008-0012-5.
- [9]. Misra, M., Kavathia, A., andSuchak, K., Int. J. Pharm. Pharma. Sci., 5(3), 2013, 371-375.
- [10]. Chaudhary, J., Jain, A., Kaur, N., and Saini, V., Int. J. Pharm. Pharm. Sci., 4(3), 2012, 191-194.
- [11]. Patel, A., Int. J. Pharm. Bio. Sci., 4(3), 2013, 379 391.
- [12]. Chaudhari, B. G., Int. J. Pharm. Chem. Biol. Sci., 2(4), 2012, 626-632.
- [13]. Patil, P., Barge, V., More, H., and Piswikar, S., Int. J. Curr. Pharm. Res., 3(2), 2011, 74-79.
- [14]. Saraf, S., Garg, G., and Saraf, S., Trend. Appl. Sci. Res., 3(3), 2008, 278-284.
- [15]. Jaina, N., Jaina, R., Jain, A., Pandeyc, S. P., and Jainc, D. K., Eurasian J. Anal. Chem., 5(3), 2010, 212-217.
- [16]. Chaudhari, B. G., Patel, A. B., Int. J. Chemtech. Res., 2(1), 2010, 633-639.
- [17]. Gupta, K. R., Mahapatra, A. D., Wadodkar, A. R., andWadodkar, S. G., Int. J. Chemtech. Res., 2(1), 551-556.

- [18]. Thomas, A. B., Jagdale, S. N., Dighe, S. B., and Nanda, R. K., Int. J. Pharmtech. Res., 2(2), 2010, 1334-1341.
- [19]. Muthu, A. K., Sankala, R., Prasad, C. S., Kumar, D. S., and Manavalan, R., Int. Res. J. Pham, 2(5), 2011, 175-180.
- [20]. Patil, P. R., Rakesh, S. U., Dhabale, P. N., andBurade, K. B., Asian J. Res. Chem., 2(1), 2009, 183-187.
- [21]. Patil, P. R., Rakesh, S. U., Dhabale, P. N. andBurade, K. B, Res. J. Pham. Technol., 2(2), 2009, 304-307.
- [22]. Rathee, P., Rathee, S., Thakur, S., and Kumar, V., Int. J. ChemTech. Res., 2(1), 2010, 62-68.
- [23]. Vichare, V., Tambe, V., Kashikar, V., and Dhole, S. N., Int. J. Chem. Res., 2(1), 7-10.
- [24]. Pawar, P. Y., Joshi, R. S., Sandhan, V., Wagh, S., and Jangale, K., Der Pharm. Lett., 3(3), 2011, 397-403.
- [25]. Amin, A., and Parikh, N., Int. J. Pharm. Res. Biosci., 1(3), 2012, 204-219.
- [26]. Baheti, K. G., and Swagatika, P., Int. J. Adv. Pharm. Biol. Sci., 2(2), 2012, 155-164.
- [27]. Kavathia, A., and Misra, M., Arabian J. Chem., 2013.
- [28]. Patel, D. B., Mehta, F. A., and Bhatt, K. K., Novel Sci. Int. J. Pharm. Sci., 1(2), 2012, 74-82.
- [29]. Lakshmi, K. S., and Lakshmi, S., J. Anal. Methods Chem., 2012, Article ID 108281, 5. doi:10.1155/2012/108281
- [30]. Vora, D. N., and Kadav, A. A., Indian J. Pharm. Sci., 70(4), 2008, 542-546.
- [31]. Altınoz, S., Celebier, M., Kaynak, M. S., and Sahin, S., Braz. J. Pharm. Sci., 46(4), 2010, 761-768.
- [32]. Islam, S. M. A., Haque, M. A., Naznin, A., Hamidul, A.N.M, and Hossain, M. K., *Dhaka Univ. J. Pharm. Sci.*, 9(2), 2010, 131-138.
- [33]. Patil, P. S., More, H. N., and Pishwikar, S. A., Int. J. Pharm. Pharm. Sci., 3(3), 2011, 146-149.
- [34]. Shabana, M. K., Chapala, D., Puttagunta, and Babu, S., Int. Res. J. Pham., 4(8), 2013, 177-183. DOI:10.7897/2230-8407.04 835.
- [35]. Baokar, S. B., Erande, R. S., and Shaikh, S. G., Indo Am. J. Pharm. Res., 2(1), 2011, 100-110.
- [36]. Tengli, A. R., Gurupadayya, B. M., and Soni, N., Int. J. Chem. Anal. Sci., 4, 2013, 33-38.
- [37]. Patil, P. R., Rakesh, S. U., Dhabale, P. N., andBurade, K. B., Int. J. ChemTech Res., 1(3), 2009, 464-469.
- [38]. El-Gizawy, S. M., Abdelmageed, O. H., Omar, M. A., Deryea, S. M., and Abdel-Megied, A. M., *Am. J. Anal. Chem.*, 3, 2012, 422-430. doi.org/10.4236/ajac.2012.36055.
- [39]. Eranki, R. J. V., Inti, G., Jayaraman, V., Vidiyala, S. R., andSreeramulu, J., ISRN Med. Chem., 2013, Article ID 178269, 1-9. doi.org/10.1155/2013/178269
- [40]. Shingare, M. S., Naidu, K. R., and Kale, U. N., J. Pharm. Biomed. Anal., 39, 2005, 147–155.
- [41]. Shah, D. A., Bhatt, J. U., Bhatt, K. K., Baldania S. L., and Chhalotiya, U. K., *Novel Sci. Int. J. Pharm. Sci.*, 1, 2012, 33-39.
- [42]. Ponnerid, V., Ravia, V. B., Inamadugu, J. K., Pilli, N. R., and Sreenivasuluc, V., J. Pharm. Anal., 2(5), 2012, 319–326.
- [43]. Shinde, D. B., Patil, K. R., Rane, V. P., Sangshetti, J. N., and Yeole, R. D., J. Chromatogr. Sci., 48, 2010, 601-606.
- [44]. Szabo, L., Chis, V., Pirnau, A., Leopold, N., Cozar, O., andOrosz, S., J. Mol. Struct., 2009, 924-926, 385-392.
- [45]. Sherazi, S. T. H., Ali, M., and Mahesar, S. A., Vib. Spectrosc. 55(1), 2011, 115-118. doi:10.1016/j.vibspec.2010.09.010
- [46]. Mahesar, S. A., Bendini, A., Cerretani, L., Carboginin, M. B., and Sherazi, S. T. H., Eur. J. Lipid Sci. Technol., 2010, vol. 112, no. 12, p. 1356-1362. doi:10.1002/ejlt.201000388
- [47]. European Pharmacopoeia, 2002.
- [48]. United States Pharmacopoeia, 2002.
- [49]. Matkovic, S. R., Valle, G. M., and Briand, L. E., Latin Am. Appl. Res., 35, 2005, 189-195.
- [50]. European Pharmacopoeia, 5, 2005.
- [51]. Qasim, B. H., J. Eng. Technol., 28(15), 2010, 1-5.
- [52]. Kenos, E., Samadizadeh, A. H., and Boreiri, S. , E-J. Chem., 9(4), 2012, 2232-2238.

- [53]. Patraa, K. C., Kumar, K. J., Sahooa, S. K., and Suresha, P., Eurasian J. Anal. Chem., 5(1), 2010, 73-80.
- [54]. Prasad, R. R., Bhuvaneswari, K., Muralilal, and Rajani, K., *Int. J. Res. Pharm. Biomed. Sci.*, 3(4), 2012, 1523-1525.
- [55]. Bhongade, B., Talath, S., andDhaneshwar, S., Int. J. Spectrosc., Article ID 294612, 2014, 1-6. doi.org/10.1155/2014/294612
- [56]. Vieira, D. C. M., Ricarte, P. C., and Salgado, H. R. N., Adv. Anal. Chem., 2(6), 2012, 80-87. DOI: 10.5923/j.aac.20120206.02.
- [57]. Sherazi, S. T. H., Mallah, M. A., Mahesar, S. A., and Khaskheli, A. R., Am. J. Anal. Chem., 3, 2012, 503-511. doi:10.4236/ajac.2012.38067.
- [58]. ICH. International Conference on Harmonization Q2 (R1), Validation of Analytical Procedures: Text and Methodology.
- [59]. ICH. International Conference on Harmonization Q1A (R2), Stability Testing of New Drug Substances and Products. 2000
- [60]. ICH. International Conference on Harmonization Q1B (R2), Photo Stability Testing on New Drug Substances and Products. 1996.