Available Online at: www.ijpar.com

[Research article]

ISSN: 2320-2831

A New Validated Stability-Indicating RP-HPLC Method for the Estimation of Pitavastatin in Tablet Dosage Forms

 $K.Sujatha^{*1}$, J.V.L.N.Seshagiri Rao^2

ABSTRACT

An accurate and stability - indicating high performance liquid chromatographic method was developed for quantification of pitavastatin in its tablet dosage forms. Ideal separation of the drug was achieved on an Agilent Eclipse XDB C_{18} column (150 x 4.6 mm; 5μ) by eluting with a mobile phase consisting of a mixture of phosphate buffer (pH 3.4) and acetonitrile (65:35 v/v) at a flow rate of 0.9 mL/min. The drug in the eluates was monitored at 244 nm. Under optimized conditions, the retention time obtained for the drug was 3.905 min. The calibration plot was linear in the concentration range of 25 - 150 μ g/mL of the drug. The validation of the method was done by following the ICH guidelines. The proposed method could be applied for determination of pitavastatin in its tablet dosage forms without any interference from normal excipients. The method thus, is suitable for routine quality control analysis of pitavastatin.

Key words: Pitavastatin, Estimation, Tablets, Stability-indicating, HPLC.

INTRODUCTION

Pitavastatin, [(3R,5S)-7-(2-cyclopropyl-4-(4-fluoro phenyl) quinolin-3-yl) - 3, 5-dihydroxy 6(E)-heptenoic acid],[1] is a member of the medication class of statins. Pitavastatin is a novel, synthetic 3-hydroxy-3-methylglutaryl coenzyme-A (HMG-CoA) reductase inhibitor [2-3], approved for the treatment of hypercholesterolemia. HMG-CoA

enzyme inhibitors inhibit the synthesis mevalonate, a rate limiting step in synthesis of cholesterol, resulting in the lowering cholesterol.[4] It is a competitive inhibitor and exerts its potent pharmacological actions by strongly binding to the active sites on HMG-CoA reductase. It has more potent lipid lowering action than other statins [5].

¹Department of Pharmacy, Government Polytechnic, Visakhapatnam, A.P., India.

²Yalamarty College of Pharmacy, Visakhapatnam, A.P. India.

^{*} Corresponding author: K.Sujatha. E-mail address: sujikandi@yahoo.in

Figure 1. Structure of pitavastatin

A literature survey revealed that some analytical methods have been reported for the determination of pitavastatin in pharmaceutical dosage forms using spectrophotometry [6], HPLC [7], HPTLC [8-9], UPLC [10] and LC/MS [11]. We have developed a

MATERIALS AND METHODS

The pure reference sample of pitavastatin was obtained from Aizant Drug Research Solutions Pvt Ltd., Hyderabad. The commercial tablet formulation of pitavastatin 'Flovas' (2 mg) manufactured by IPCA Laboratories Ltd., Mumbai, was purchased from the local market. Potasium dihydrogen

B. Equipment and chromatographic conditions

A Waters Alliance liquid chromatograph (model 2695) fitted with a diode array detector (model 2996) and running on Empower2 data handling system was employed in the study. An Agilent Eclipse XDB C_{18} column (150 x 4.6 mm; 5μ) was used for analyzing the drug. All the chromatographic runs were carried out by using a mobile phase consisting of a mixture

new accurate and precise stability – indicating RP-HPLC method with short retention and run times for the determination of pitavastatin in bulk drug samples and in tablet dosage forms. The developed method has been duly validated as per ICH guideline.

A. Drugs, chemicals, and solvents

orthophosphate, orthophosphoric acid, HPLC grade acetonitrile and HPLC grade methanol were purchased from Rankem Fine Chemicals Ltd., Mumbai. HPLC grade water was prepared by using Millipore Milli-Q system.

of phosphate buffer (pH 3.4) and acetonitrile (65:35 v/v) in isocratic mode at a flow rate of 0.9 mL/min. The injection volume of the samples was 10 μ L. The detector wavelength was set at 244 nm. The chromatographic run time was set as 8.0 min. Under these optimized conditions, the retention time obtained for pitavastatin was 3.905 min.

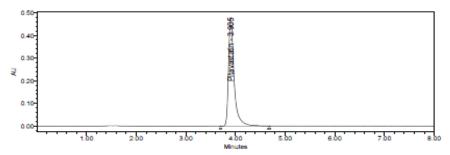


Figure: 2.Chromatogram of Pitavastatin from working standard solution

C. Preparation of the phosphate buffer

The phosphate buffer was prepared by dissolving 1.36 gm of potassium dihyrogen phosphate in a beaker containing 1000 mL of water and the contents

were sonicated. The pH of the solution was then adjusted to 3.4 with ortho phosphoric acid. It was then filtered through a 0.45μ membrane filter.

D. Preparation of the mobile phase

The optimized mobile phase consisted of a mixture of the above-mentioned phosphate buffer (pH 3.4) and acetonitrile in the ratio of 65:35 v/v.

E. Preparation of the diluent

Methanol was used as the diluent.

F. Preparation of the working standard solution of pitavastatin

10 mg of pitavastatin reference standard was accurately weighed and transferred into a 10 mL volumetric flask. To this, 7 mL of methanol was added, sonicated for 5 minutes and the volume was made up with a further quantity of methanol. This was used as the standard stock solution. The working standard solution was prepared by diluting 1.0 mL of

H. Estimation of the drug from the tablet dosage forms

Ten 'Flovas' (2 mg) tablets were crushed and ground to a fine powder. Tablet powder equivalent to 10 mg of pitavastatin was transferred into a 10 mL volumetric flask. 8mL of diluent was added and sonicated for 30 min. The volume was made up with the diluent and the contents were mixed well. This mixture was filtered through a 0.45μ membrane filter

the standard stock solution to 10 mL with the diluent in a volumetric flask.

G. Calibration plot

Solutions of pitavastatin at different concentration levels including the working standard concentration were prepared with the diluent. Twenty microlitres of each concentration was injected three times into the HPLC system (n=3). The response was read at 244 nm and the corresponding chromatograms were recorded. From these chromatograms, the mean peak areas at the different concentration levels were calculated and the linearity plot of mean peak areas over their concentrations was constructed.

(discarding the first few mL of the filtrate). 1 mL of the filtrate was transferred into a 10 mL volumetric flask and made up to volume with diluent. This solution was then chromatographed six times. From the chromatograms obtained, the average drug content in the formulation was calculated.

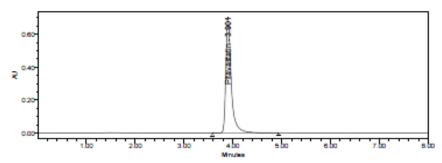


Figure 3. Chromatogram of pitavastatin from its tablet dosage form

RESULTS AND DISCUSSION

During the method optimization studies, various combinations and proportions of the solvents and buffers were examined on an Agilent Eclipse XDB C_{18} column for efficient separation of pitavastatin. Using a mobile phase consisting of a mixture of phosphate buffer (pH 3.4) and acetonitrile in the ratio of 65:35 v/v, a good resolution and baseline separation of the drug peak was obtained. All the chromatographic conditions were optimized by

valuating the column efficiency parameters like theoretical plates and tailing factor (Table 1). Under these optimized conditions, the retention time obtained for pitavastatin was 3.905 min (Figure 2) in a run time of 8.0 min. The method was then validated as per the ICH guideline. The proposed method was also found to be applicable for the analysis of pitavastatin in tablet formulations.

Table 1. Optimized chromatographic conditions

Stationary Phase	Agilent Eclipse XDB C18 (100 x 4.6 mm, 5μm)		
Mobile Phase	Phosphate buffer : Acetonitrile =65:35 v/v		
Flow Rate	0.9 mL/min		
Column Temperature	30°C		
Injection Volume	10 μL		
Detection Wavelength	244 nm		

A. Specificity

A good analytical method should be able to measure the analytes accurately in the presence of probable interferences from its solvent as well as from the excipients of its formulation. Figure 2 shows good chromatographic baseline separation of pitavastatin from its working standard solution. Figure 3 demonstrates that no interfering peaks were observed at the retention time of pitavastatin arising due to the excipients of its tablet.

B. Linearity

The calibration curve (n=3) constructed for the drug was linear over the concentration range of 25-150 µg/mL. The regression of the plot was computed by least square regression method and is shown in

Figure 4. The correlation coefficient is greater than 0.99 and the %RSD at each concentration studied was less than 2.

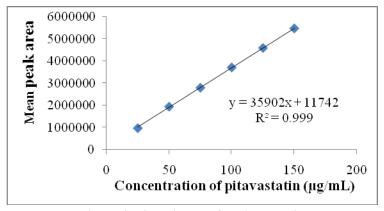


Figure 4. Linearity plot for pitavastatin

C. Accuracy and precision

The accuracy of the method was determined by recovery experiments. The recovery studies were carried out and the percent recovery with its standard deviation were calculated (Table 2). The high percentage of recovery indicates that the proposed

method is quite accurate. The precision of the method was demonstrated by inter-day and intra-day variation studies. Six replicate injections of sample solutions were made and the percent RSD was calculated (Table 3).

Table 2. Accuracy data of the proposed method

Amount of the analyte	Mean recovery	% Mean recovery ±
taken (µg/mL)	$(\mu g/mL) \pm SD$	SD
50	50.26 ± 0.31	100.52 ± 0.62
100	100.14 ± 0.01	100.14 ± 0.01
150	151.65 ± 0.03	101.10 ± 0.02
	taken (μg/mL) 50 100	50 50.26 ± 0.31 100 100.14 ± 0.01

rubic of recession data for the proposed method				
S.No.	Intra-day precision	Inter-day precision		
1	3722681	3721986		
2	3722486	3722045		
3	3722687	3718926		
4	3722415	3722694		
5	3722105	3722468		
6	3722098	3722195		
Average	3722412	3721719		
SD	263.18	1394.03		
%RSD	0.007	0.037		

Table 3. Precision data for the proposed method

D. System suitability parameters

System suitability parameters were studied with six replicate injections of the standard solution and the results are presented in Table 4.

Table 4. System suitability parameters of the proposed method

Parameter	Value
Retention time (min)	3.905
Tailing factor	1.1
Theoretical plates	7955
HETP	0.0126

E. Degradation studies

Peroxide degradation: 1 mL of stock solution of pitavastatin was transferred into a 10 mL volumetric flask. To that 1 mL of 20% hydrogen peroxide (H₂O₂) was added. The solution was kept for 30 min at 60°C. The resultant solution was

diluted with diluent to obtain 100 $\mu g/mL$ solution of pitavastatin. 10 μL of this solution was injected into the system and the chromatogram was recorded.

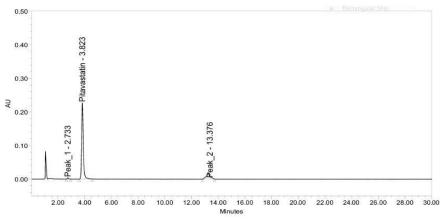


Figure 5. Chromatogram of pitavastatin subjected to peroxide degradation

Acid - degradation: 1 mL of stock solution of pitavastatin was transferred into a 10 mL volumetric flask. To it 1 mL of 2M hydrochloric

acid was added and refluxed for 30mins at 60°C. The resultant solution was diluted with diluent to obtain 100 µg/mL solution of

pitavastatin. 10 µL of this solution was injected into

the system and the chromatogram was recorded.

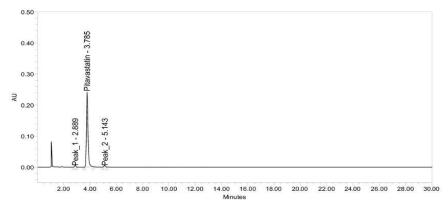


Figure 6. Chromatogram of pitavastatin subjected to acid - degradation

Base - degradation: 1 mL of stock solution of pitavastatin was transferred into a 10 mL volumetric flask. To it, 1 mL of 2 M sodium hydroxide was added and refluxed for 30min at

 $60^{\circ}C.$ The resultant solution was diluted with diluent to obtain 100 $\mu g/mL$ solution of pitavastatin. 10 μL of this solution was injected into the system and the chromatogram was recorded.

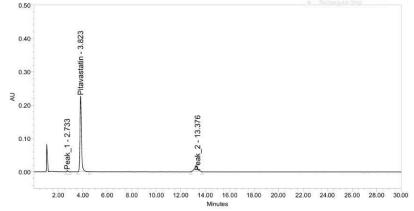


Figure 7. Chromatogram of pitavastatin subjected to base - degradation

Dry heat degradation: The working standard solution of pitvastatin was placed in oven at 105°C for six hours. The resultant solution was

diluted to 100 $\mu g/mL$ solution and 10 μL was injected into the system and the chromatogram was recorded.

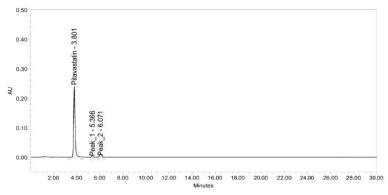


Figure 8. Chromatogram of pitavastatin subjected to dry heat - degradation

Photo - degradation: The photochemical stability of the drug was also studied by exposing the working standard solution of pitavastatin to UV light by keeping the beaker in UV Chamber for 7days.

The resultant solution was diluted to obtain 100 $\mu g/mL$ solution of pitavastatin. 10 μL of this solution was injected into the system and the chromatogram was recorded.

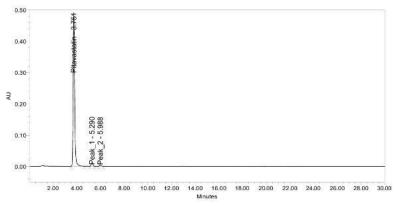


Figure: 9. Chromatogram of pitavastatin subjected to photo - degradation

Neutral - degradation: Stress testing under neutral conditions was studied by refluxing the drug in water for 6 hours at a temperature of 60°C. The

resultant solution was diluted to get $100~\mu g/mL$ solution of pitavastatin and $10~\mu L$ was injected into the system and the chromatogramswas recorded.

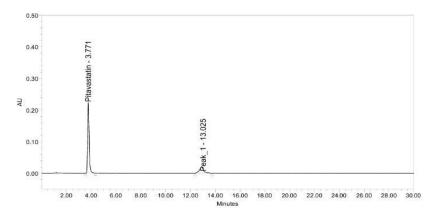


Figure 10. Chromatogram of pitavastatin subjected to neutral - degradation

F. Method suitability

The commercial tablet formulation, Flovas (2 mg) was analyzed by the proposed method. The recovery obtained (100.1%) by the proposed method was found to be in good agreement with the labelled

amount of the drug, which confirms the suitability of the method for the analysis of pitavastatin in tablet dosage forms.

CONCLUSION

The proposed RP -HPLC method is sensitive, precise, accurate and stability indicating and can be used for

the routine determination of pitavastatin in its tablet dosage forms.

ACKNOWLEDGEMENT

The authors are thankful to the authorities of the College of Pharmaceutical Sciences, Andhra

University, for providing laboratory facilities to carry out this study.

REFERENCES

- [1] Sweetman SC, Martindale. The complete dug reference. The Pharmaceutical press: London, 2007.
- [2] Kajinami K, Takekoshi N, Saito Y. Pitavastatin: Efficacy and Safety Profiles of a Novel Synthetic HMG CoA Reductase Inhibitor. *Cardiovascular Drug Reviews*. 21, 2003, 199 215.
- [3] Mukhtar RYA, Reid J, Reckless JPD. Pitavastatin, International Journal of Clinical Practice. 59, 2005, 239 252.
- [4] Lennernäs H, Fager G. Pharmacodynamics and pharmacokinetics of the HMG-CoA reductase inhibitors-Similarities and differences. *Clinical Pharmacokinetics*. 32, 1997, 403 – 425.
- [5] Istvan ES, Deisenhofer J. Structural mechanism for statin inhibition of HMG-CoA reductase. *Science*. 292, 2001, 1160 –1164.
- [6] Krishna MV, Sankar DG. Adaptation of Color Reactions for Spectrophotometric Determination of Pitavastatin Calcium in Bulk Drugs and in Pharmaceutical Formulations. *E J Chem.* 4, 2007, 272-278.
- [7] Neelima B, Ravi Kumar P, Hima Bindu V, Rajendra Prasad Y. A Validated Stability Indicating RP-HPLC Method for Estimation of Pitavastatin in Bulk and Pharmaceutical Dosage Form. *International Journal of Pharma Sciences*. 3, 2013, 309-315.
- [8] Kumar NS, Nisha N, Nirmal J, Sonali N, Bagyalakshmi J. HPLC Determination of Pitavastatin Calcium in Pharmaceutical Dosage Forms. *Pharm Ana Acta*. 2, 2011, 119. doi:10.4172/2153-2435.1000119.
- [9] Satheesh Kumar N, Baghyalakshmi J. Determination and Quantification of Pitavastatin Calcium in Tablet Dosage Formulation by HPTLC Method. *Analytical Letters*. 40, 2007, 2625-2632.
- [10] Hiral Panchal J, Bhanubhai Suhagia N, Natubhai Patel J, Bhavesh Patel H. A Simple and Sensitive HPTLC Method for Quantitative Analysis of Pitavastatin Calcium in Tablets. *Journal of Planar Chromatography – Modern TLC*. 21, 2008, 267-270.
- [11] Antony Raj Gomas, Pannala Raghu Ram, Nimmakayala Srinivas, Jadi Sriramulu. Degradation Pathway for Pitavastatin Calcium by Validated Stability Indicating UPLC Method. American Journal of Analytical Chemistry. 2, 2010, 83-90.
- [12] Nirogi R, Mudigonda K, Kandikere V. Chromatography–Mass Spectrometry Methods for the Quantitation of Statins in Biological Samples. *Journal of Pharmaceutical and Biomedical Analysis*. 44, 2007, 379-387.
- [13] ICH Harmonized Tripartite Guidelines (Q2R1). Validation of analytical procedures: Text and Methodology. International Conference on Harmonization, European commission, Japan and USA (2005).
